CIC **Tuning wetting-dewetting thermomechanical energy** energi for hydrophobic nanopores: Preferential intrusion¹ GUNE

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^aCentre for Cooperative Research on Alternative Energies (CIC energiGUNE), Basque Research and Technology Alliance (BRTA), Alava Technology Park, Albert Einstein 48, 01510 Vitoria-Gasteiz, Spain

<u>Luis Bartolomé</u>^a, Argyrios Anagnostopoulos^b, Alexander R. Lowe^b, Piotr Ślęczkowski^b, Eder

Amayuelas^a, Andrea Le Donne^c, Michał Wasiak^d, Mirosław Chorażewski^b, Simone Meloni^c,

Yaroslav Grosu^a

^bInstitute of Chemistry, University of Silesia, 40-006 Katowice, Poland

OF LODZ UNIVERSITY OF SILESIA ^cDipartimento di Scienze Chimiche e Farmaceutiche Università degli Studi di Ferrara, Via Luigi Borsari 46, I-44121 Ferrara, Italy IN KATOWICE ^dDepartment of Physical Chemistry, Faculty of Chemistry, University of Łódź, Pomorska 165, 90-236 Łódź, Poland

Introduction

The thermomechanical energy of compression/decompression, which is relevant to many industrial and natural processes, is a prominent property of thermodynamic systems². Unfortunately, this property is challenging to tune due to fundamental limitations for simple fluids³. However, in this presentation, we demonstrate via direct experimental and atomistic observations that these fundamental limitations can be overcome by exploiting the preferential intrusion of water from aqueous solutions into sub-nanometer pores. We suggest this genuinely sub-nanoscale phenomenon has the potential to develop into a strategy for controlling the thermomechanical energy of microporous liquids as well as for tuning the heat of wetting/dewetting of nanopores. This effective control would be relevant to a great variety of natural and technological processes such as the separation of liquids, liquid-phase chromatography, porosimetry, energy dissipation, conversion and storage, biological and bioinspired channels and many more^{4,5}.

$Exo \leftrightarrow Endo$ 00 0 0 $\bigcirc \bigcirc \bigcirc \bigcirc$





the previous pressure point depending on the pressure for the three simulated alternatives. c) The intrusion heat for the three simulated alternatives i.e., water, free-KBr and restricted-KBr atoms.



· Experimental and theoretical results show that both mechanical and thermal energies of the intrusion-extrusion process can be tuned via preferential intrusion.



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- By varying the concentration of solution, one can achieve an endothermic or exothermic compression/decompression process.
- With further development, results suggest that preferential intrusion of species from a solution into sub-nanoporous materials can become a strategy for tuning the heat of the intrusion-extrusion process relevant to technological and natural systems.



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